

## Multigram-Scale Access to 2-Oxaadamantan-1-amine and 2-Oxaadamantan-1-ol via Optimized Synthesis of Bicyclo[3.3.1]nonane-3,7-dione

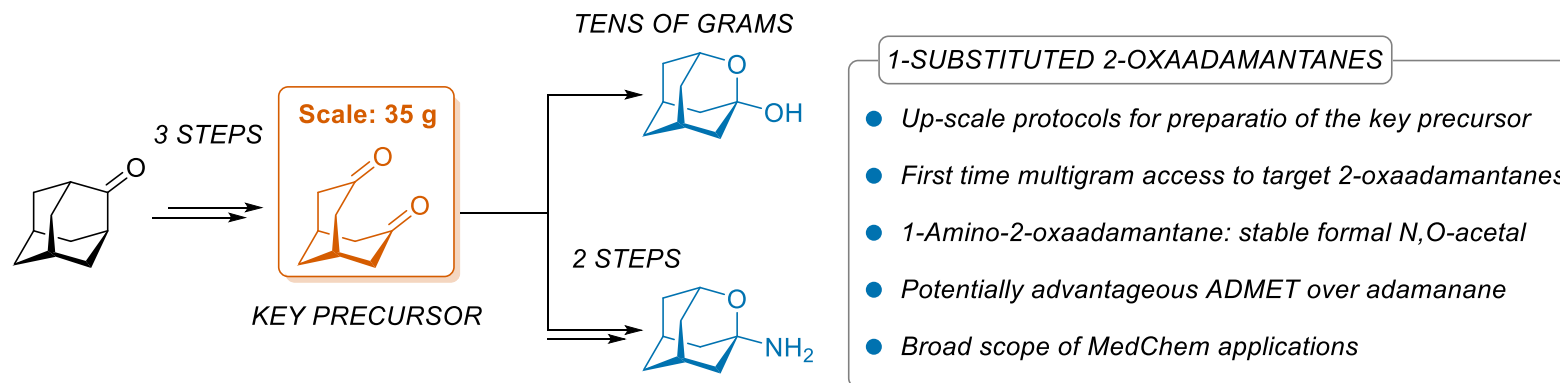
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## X-ray experimental part

The colourless crystals of compound **3** (C<sub>10</sub>H<sub>18</sub>O<sub>2</sub>) are monoclinic. At 173 K  $a = 13.081(9)$ ,  $b = 17.450(16)$ ,  $c = 10.354(6)$  Å,  $\beta = 128.81(4)^\circ$ ,  $V = 1842(3)$  Å<sup>3</sup>,  $M_r = 170.24$ ,  $Z = 8$ , space group  $C2/c$ ,  $d_{\text{calc}} = 1.228$  g/cm<sup>3</sup>,  $\mu(\text{MoK}\alpha) = 0.083$  mm<sup>-1</sup>,  $F(000) = 752$ . Intensities of 10135 reflections (1619 independent,  $R_{\text{int}} = 0.1214$ ) were measured on the Bruker APEX II diffractometer (graphite monochromated MoK $\alpha$  radiation, CCD detector,  $\varphi$ - and  $\omega$ -scanning,  $2\Theta_{\text{max}} = 50^\circ$ ). The structure was solved by direct method using OLEX2 [1] package with SHELXT [2] and SHELXL modules [3]. Positions of the hydrogen atoms were located from electron density difference maps and refined using “riding” model with  $U_{\text{iso}} = nU_{\text{eq}}$  of the carrier atom ( $n = 1.5$  for hydroxyl groups and  $n = 1.2$  for other hydrogen atoms). Full-matrix least-squares refinement against  $F^2$  in anisotropic approximation for non-hydrogen atoms using 1619 reflections was converged to  $wR_2 = 0.2847$  ( $R_1 = 0.0844$  for 650 reflections with  $F > 4\sigma(F)$ ,  $S = 0.919$ ). The final atomic coordinates, and crystallographic data for molecule **3** have been deposited to with the Cambridge Crystallographic Data Centre, 12 Union Road, CB2 1EZ, UK (fax: +44-1223-336033; e-mail: deposit@ccdc.cam.ac.uk) and are available on request quoting the deposition numbers CCDC 2543739).

The yellow crystals of compound **7** (C<sub>9</sub>H<sub>16</sub>NOCl) are monoclinic. At 173 K  $a = 6.2648(2)$ ,  $b = 15.0005(4)$ ,  $c = 9.8399(3)$  Å,  $\beta = 93.320(2)^\circ$ ,  $V = 923.15(5)$  Å<sup>3</sup>,  $M_r = 189.68$ ,  $Z = 4$ , space group  $P2_1/n$ ,  $d_{\text{calc}} = 1.365$  g/cm<sup>3</sup>,  $\mu(\text{MoK}\alpha) = 0.366$  mm<sup>-1</sup>,  $F(000) = 408$ . Intensities of 6367 reflections (1633 independent,  $R_{\text{int}} = 0.0261$ ) were measured on the Bruker APEX II diffractometer (graphite monochromated MoK $\alpha$  radiation, CCD detector,  $\varphi$ - and  $\omega$ -scanning,  $2\Theta_{\text{max}} = 50^\circ$ ). The structure was solved by direct method using OLEX2 [1] package with SHELXT [2] and SHELXL modules [3]. Positions of the hydrogen atoms were located from electron density difference maps and refined using “riding” model with  $U_{\text{iso}} = nU_{\text{eq}}$  of the carrier atom ( $n = 1.5$  for protonated amino group and  $n = 1.2$  for other hydrogen atoms). Full-matrix least-squares refinement against  $F^2$  in anisotropic approximation for non-hydrogen atoms using 1633 reflections was converged to  $wR_2 = 0.086$  ( $R_1 = 0.0329$  for 1444 reflections with  $F > 4\sigma(F)$ ,  $S = 1.088$ ). The final atomic coordinates, and crystallographic data for molecule **7** have been deposited to with the Cambridge Crystallographic Data Centre, 12 Union Road, CB2

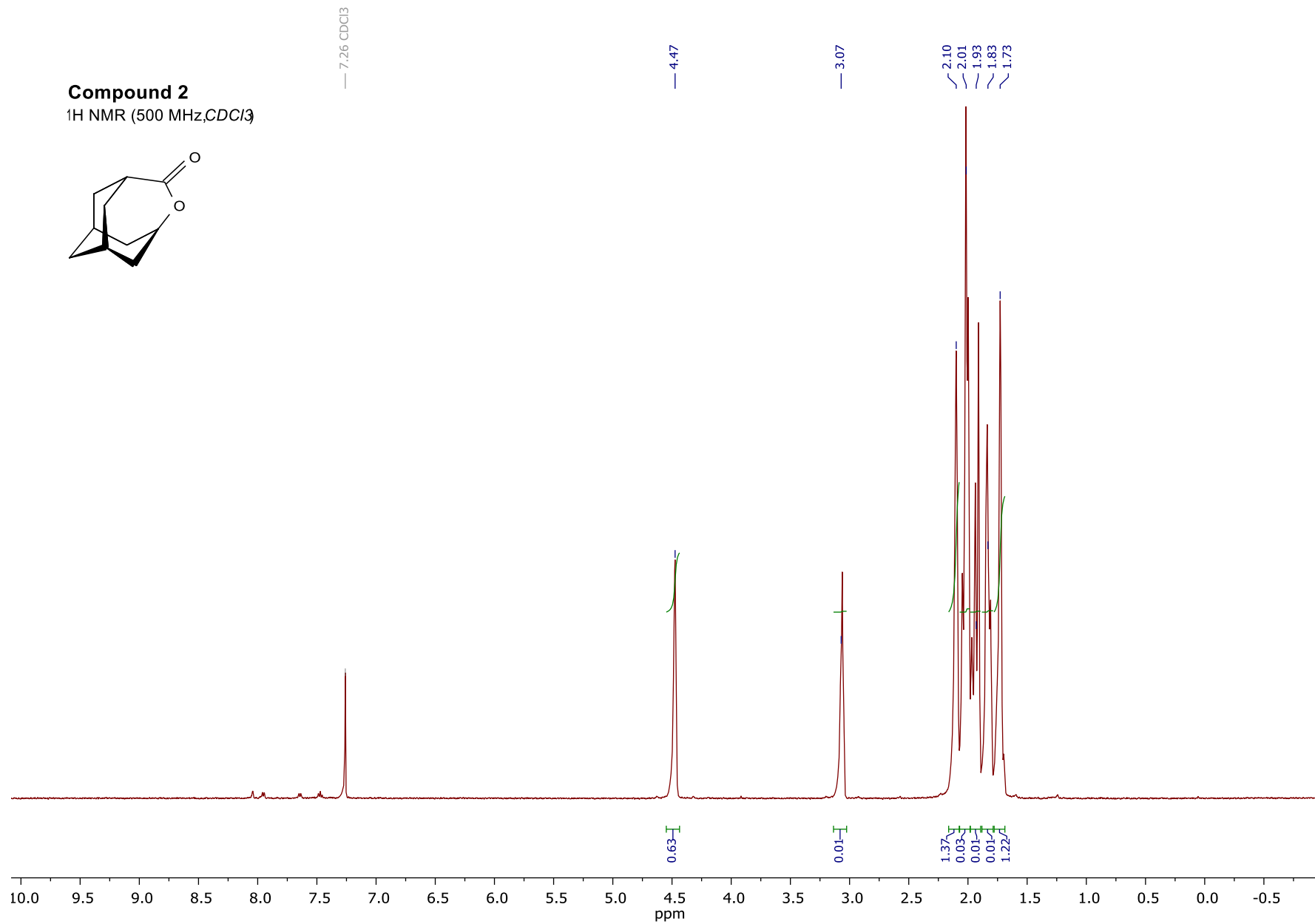
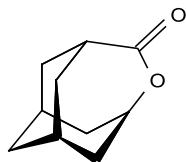
1EZ, UK (fax: +44-1223-336033; e-mail: deposit@ccdc.cam.ac.uk) and are available on request quoting the deposition numbers CCDC 2543738).

1. Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. *OLEX2*: A Complete Structure Solution, Refinement and Analysis Program. *J. Appl. Crystallogr.* **2009**, *42* (2), 339–341. <https://doi.org/10.1107/S0021889808042726>.
2. Sheldrick, G. M. *SHELXT* – Integrated Space-Group and Crystal-Structure Determination. *Acta Crystallogr., Sect. A: Found. Adv.* **2015**, *71* (1), 3–8. <https://doi.org/10.1107/S2053273314026370>.
3. Sheldrick, G. M. Crystal Structure Refinement with *SHELXL*. *Acta Crystallogr., Sect. C: Struct. Chem.* **2015**, *71* (1), 3–8. <https://doi.org/10.1107/S2053229614024218>.

# Copies of NMR spectra of the prepared compounds

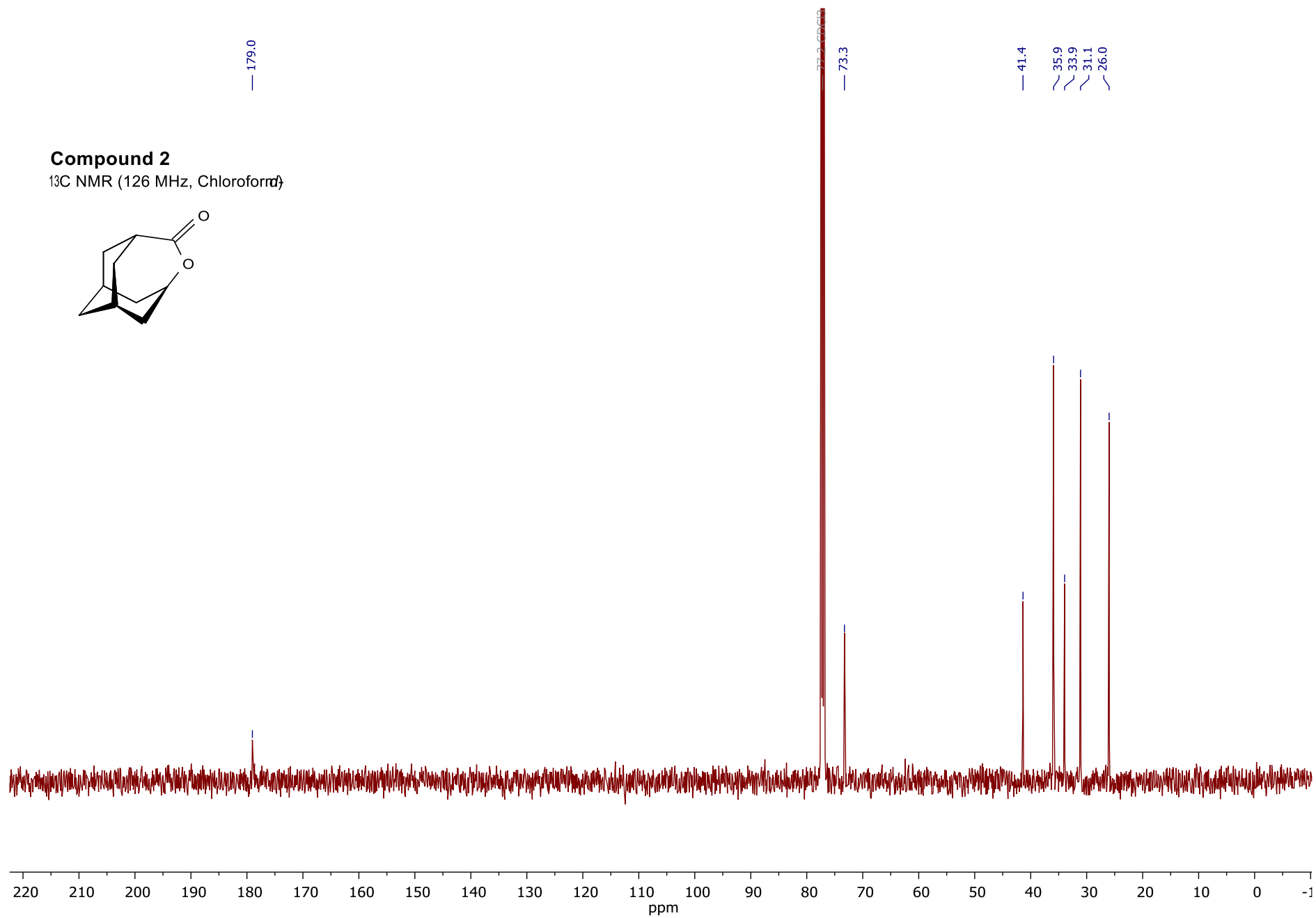
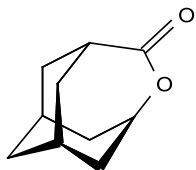
## Compound 2

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)



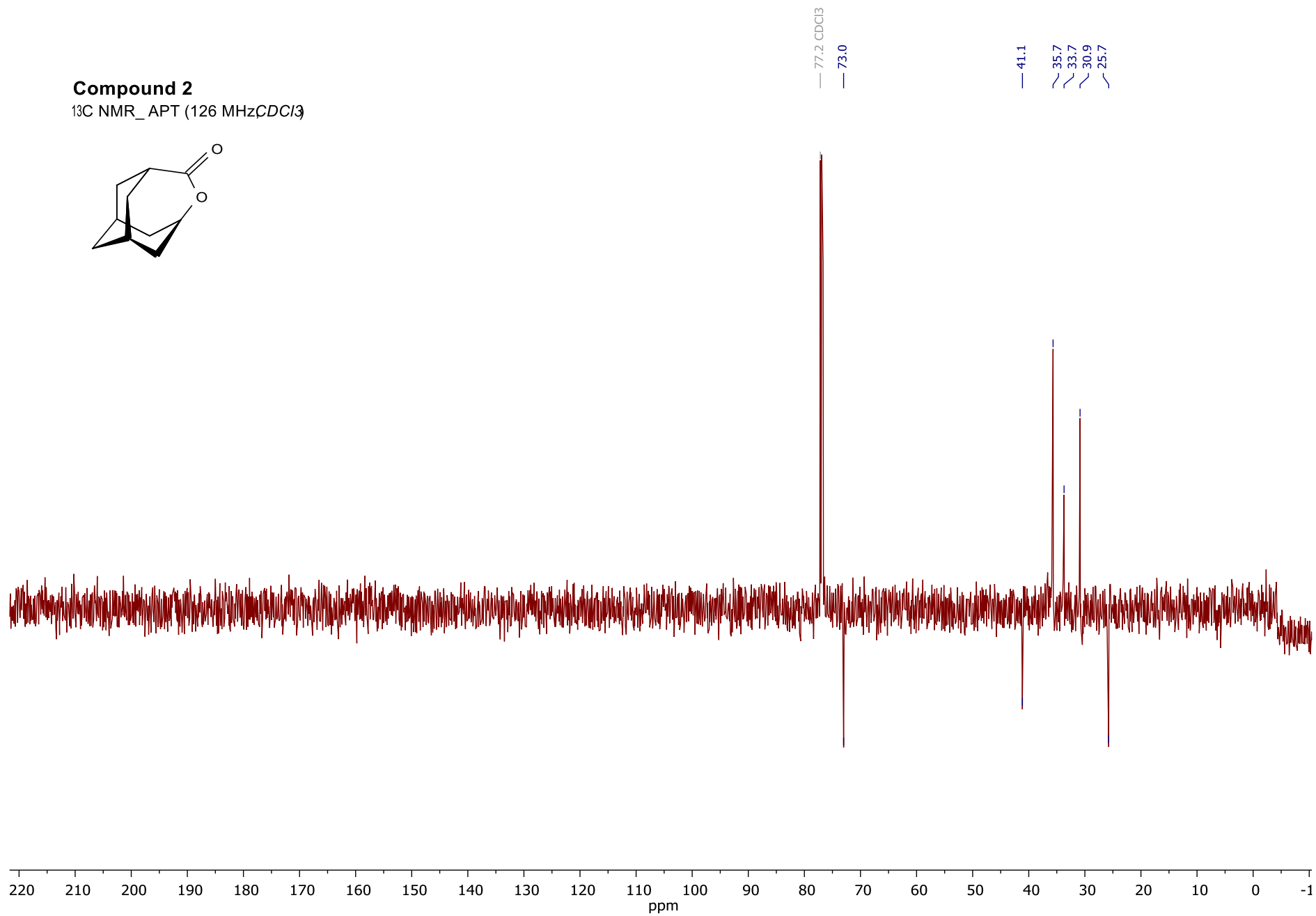
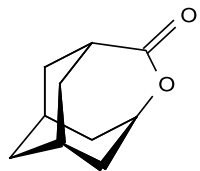
**Compound 2**

$^{13}\text{C}$  NMR (126 MHz, Chloroform)



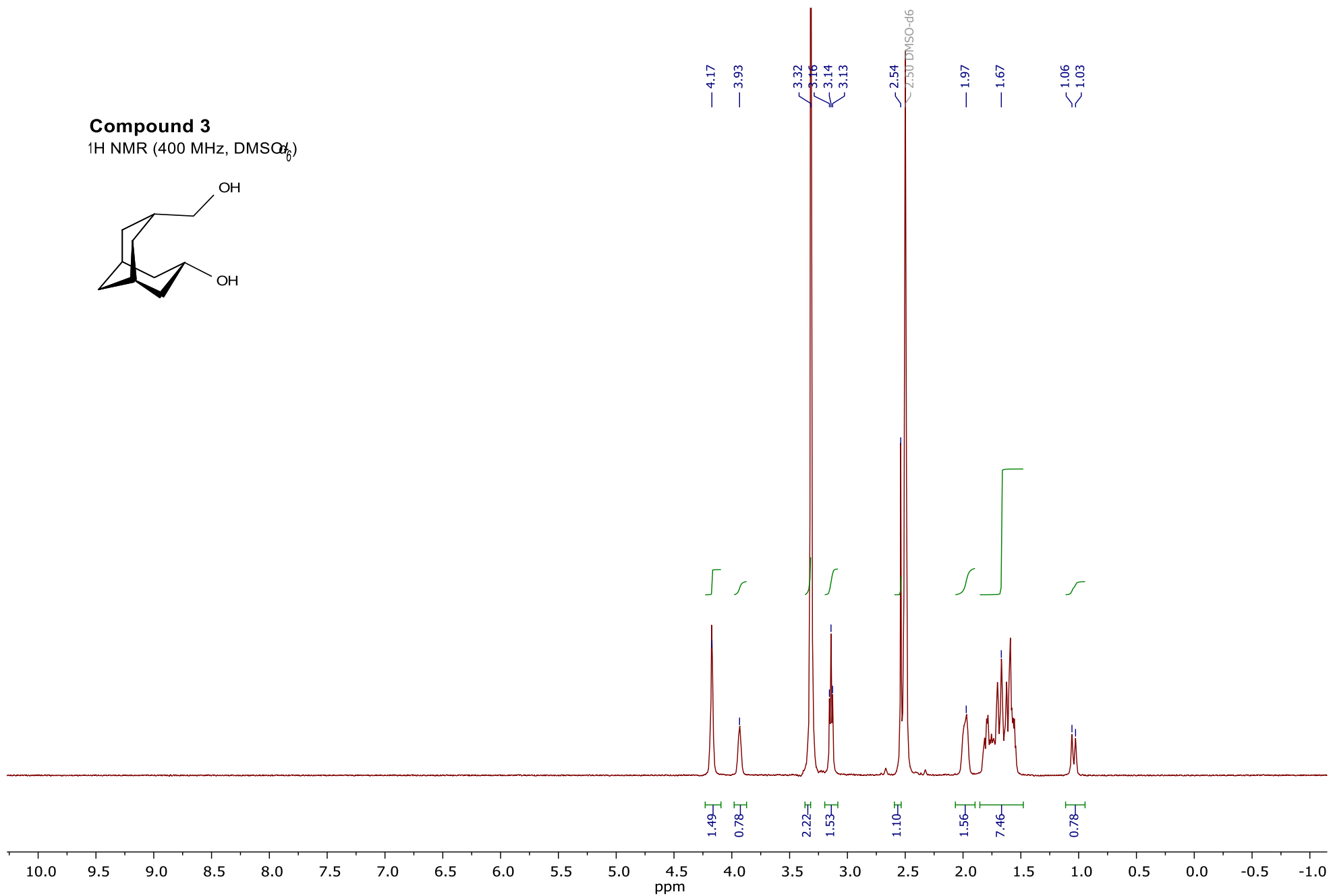
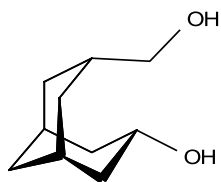
**Compound 2**

$^{13}\text{C}$  NMR\_APT (126 MHz  $\text{CDCl}_3$ )



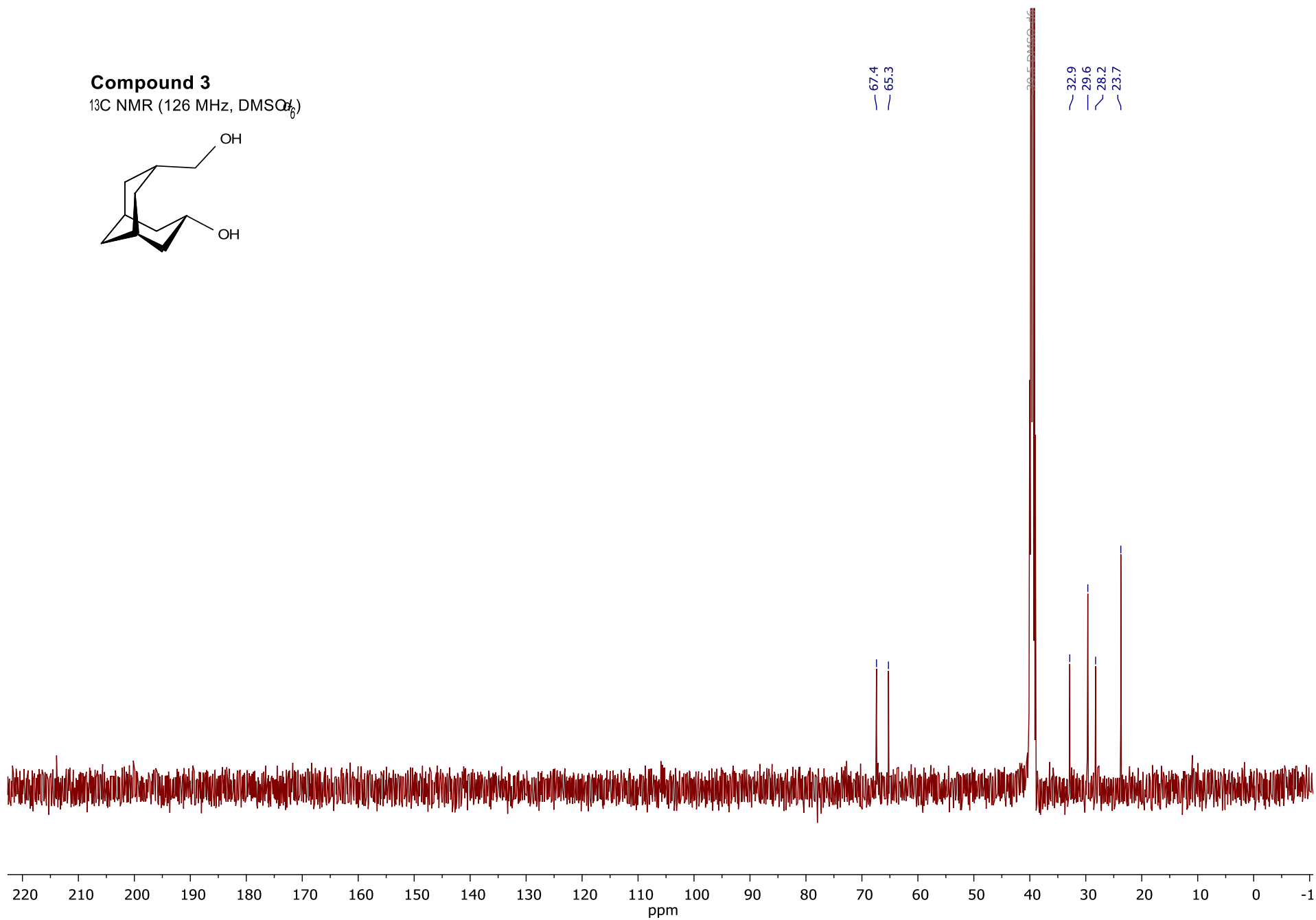
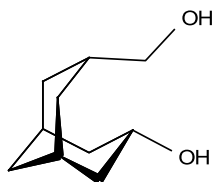
**Compound 3**

<sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>)



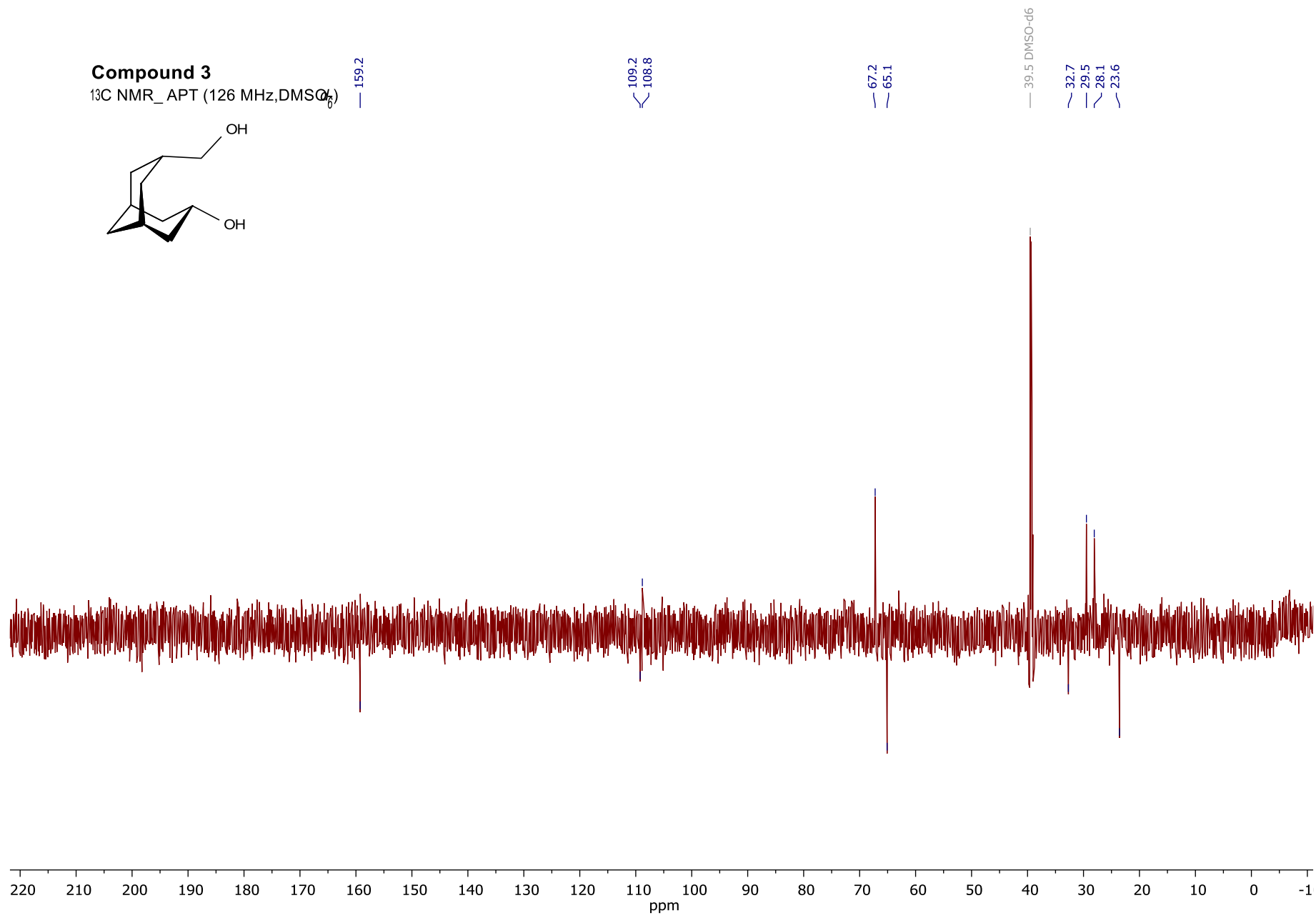
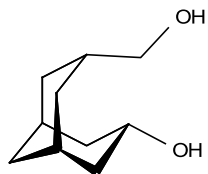
**Compound 3**

$^{13}\text{C}$  NMR (126 MHz,  $\text{DMSO-d}_6$ )

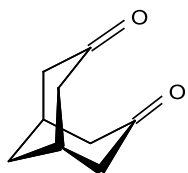


**Compound 3**

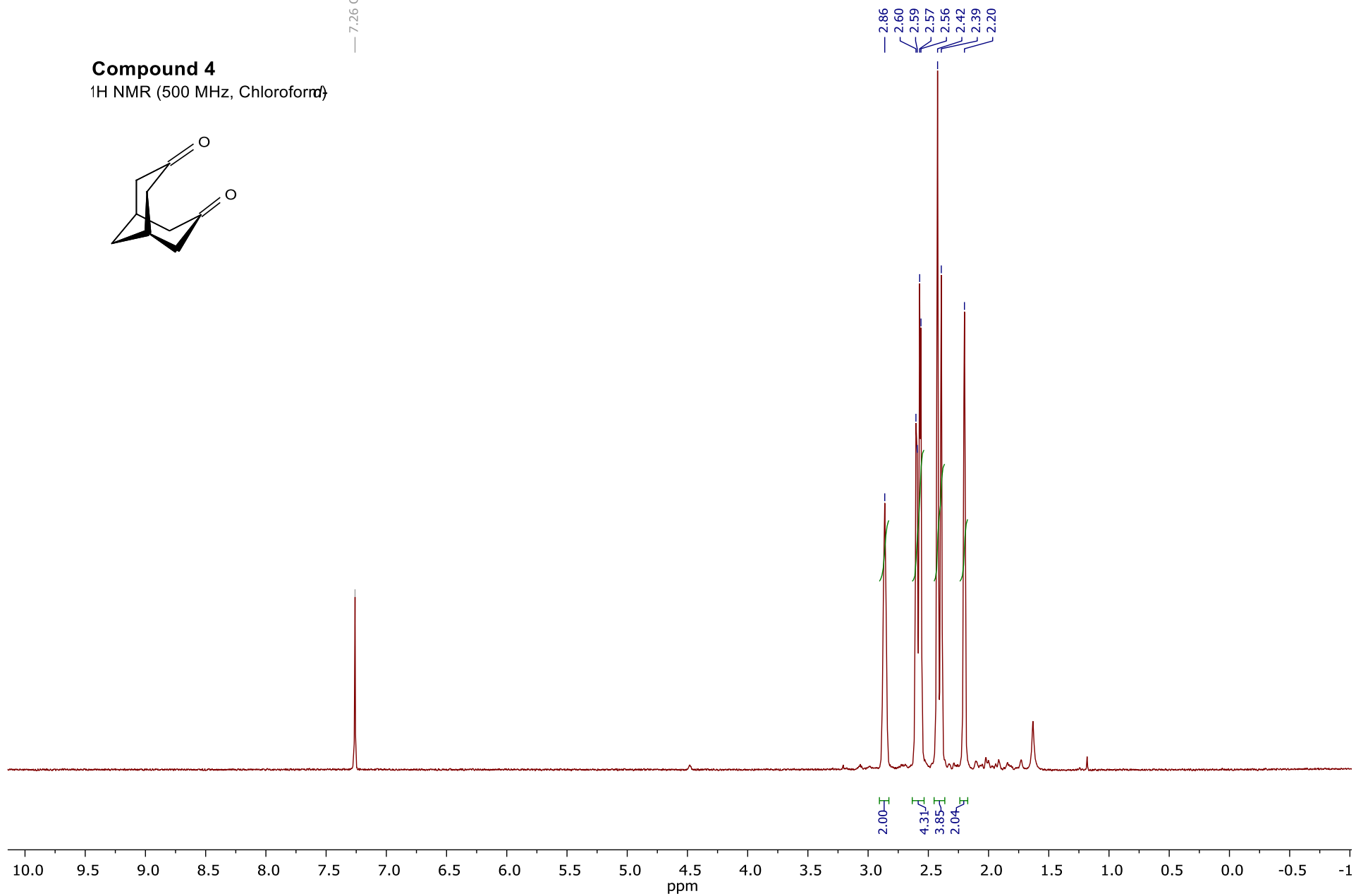
$^{13}\text{C}$  NMR\_APT (126 MHz, DMSO-d<sub>6</sub>)



**Compound 4**  
1H NMR (500 MHz, Chloroform-d)



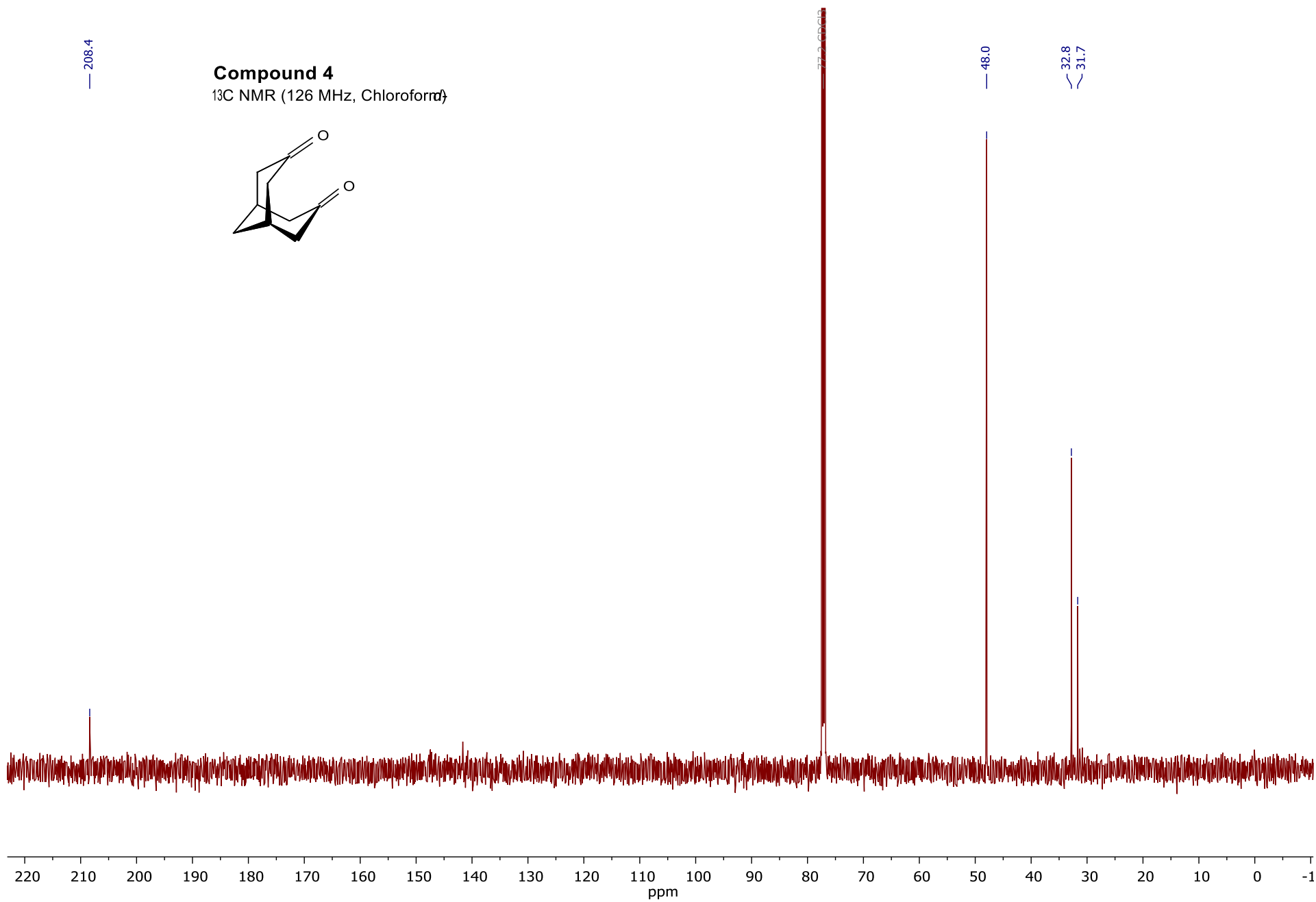
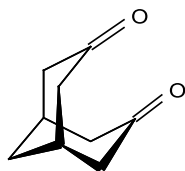
— 7.26 CDCl<sub>3</sub>



— 208.4

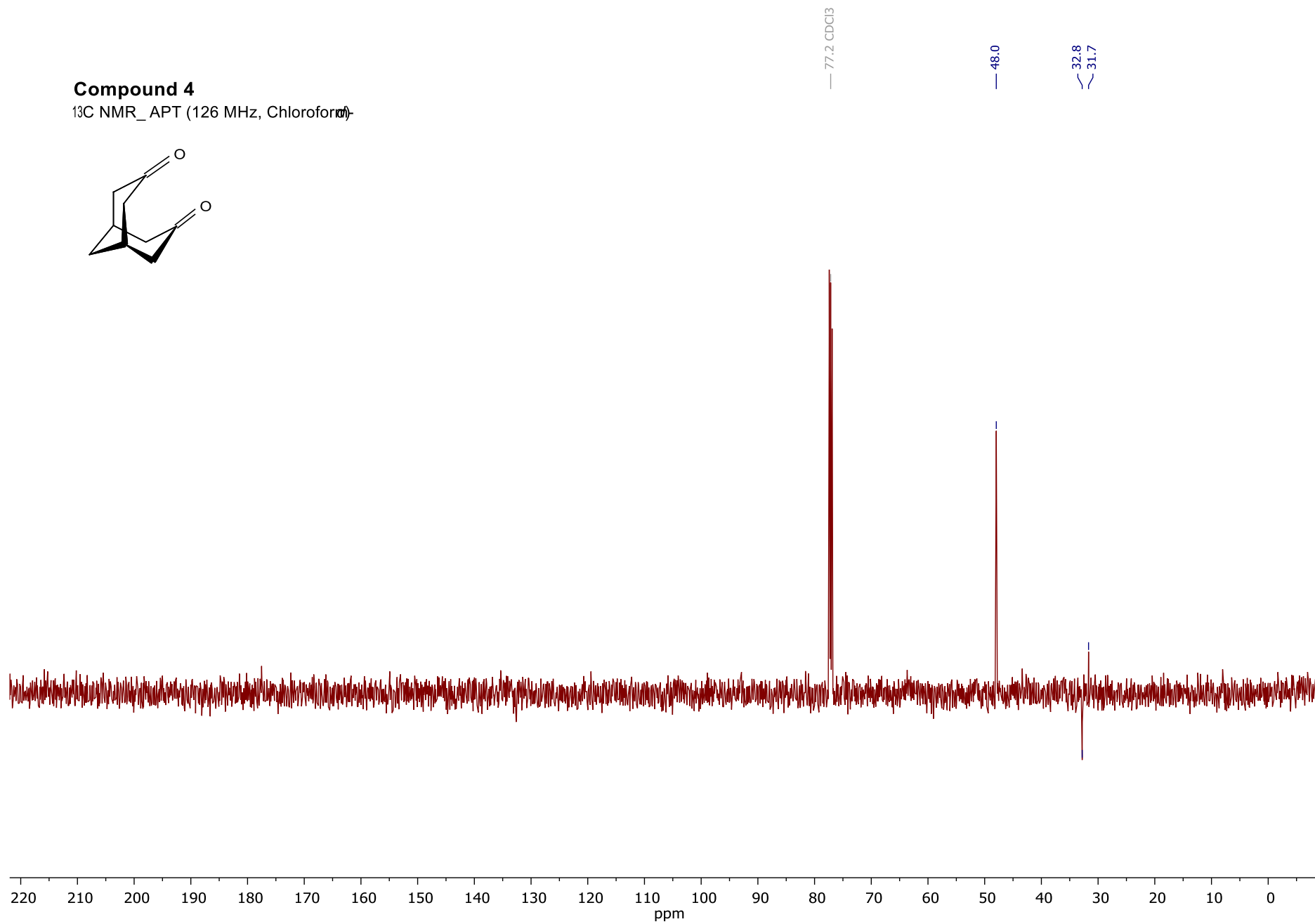
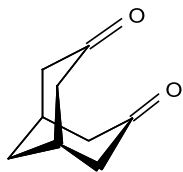
### Compound 4

<sup>13</sup>C NMR (126 MHz, Chloroform)



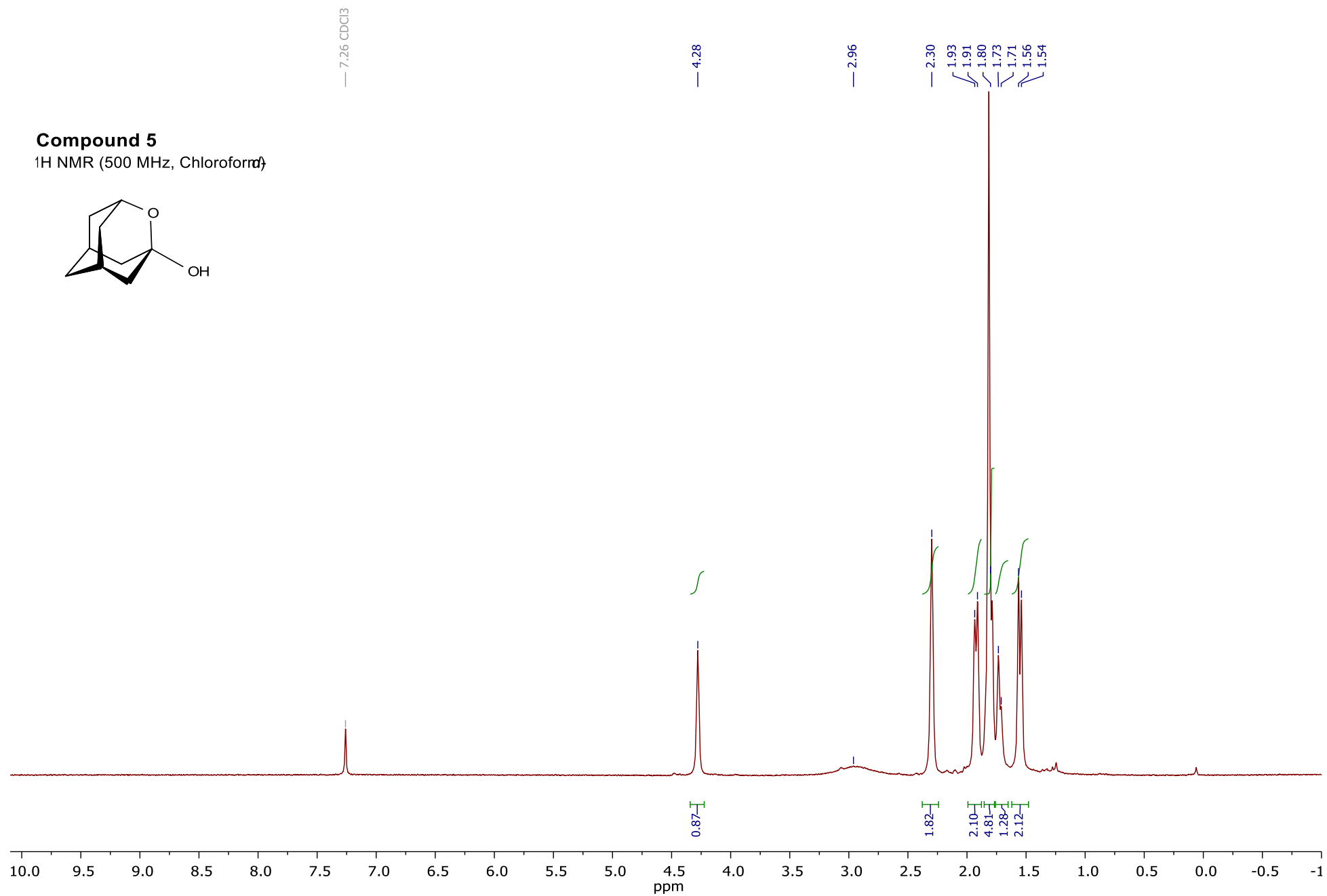
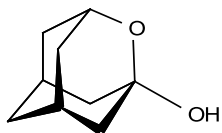
**Compound 4**

<sup>13</sup>C NMR\_APT (126 MHz, Chloroform)



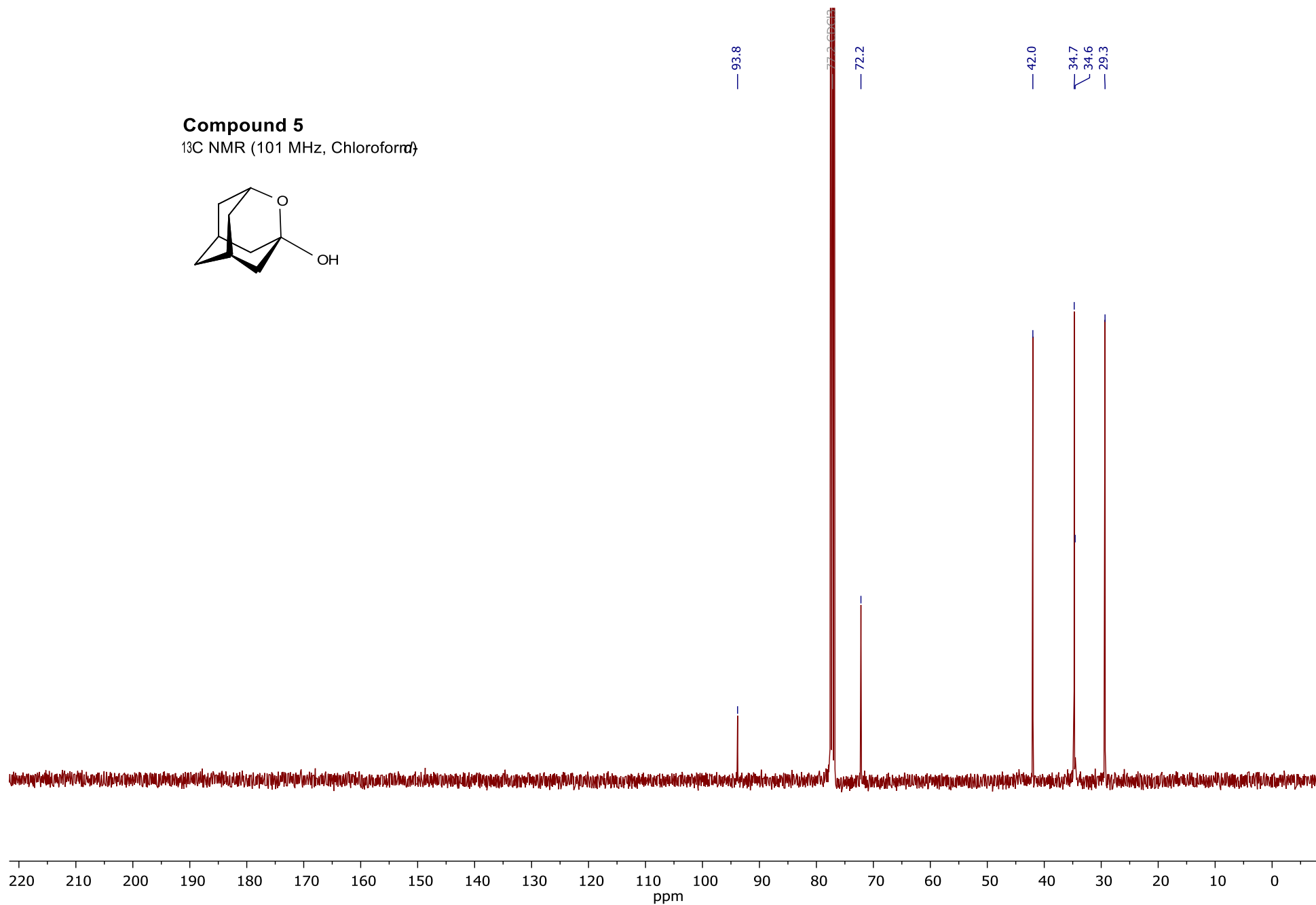
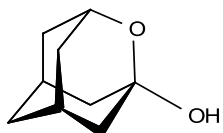
**Compound 5**

<sup>1</sup>H NMR (500 MHz, Chloroform-d)



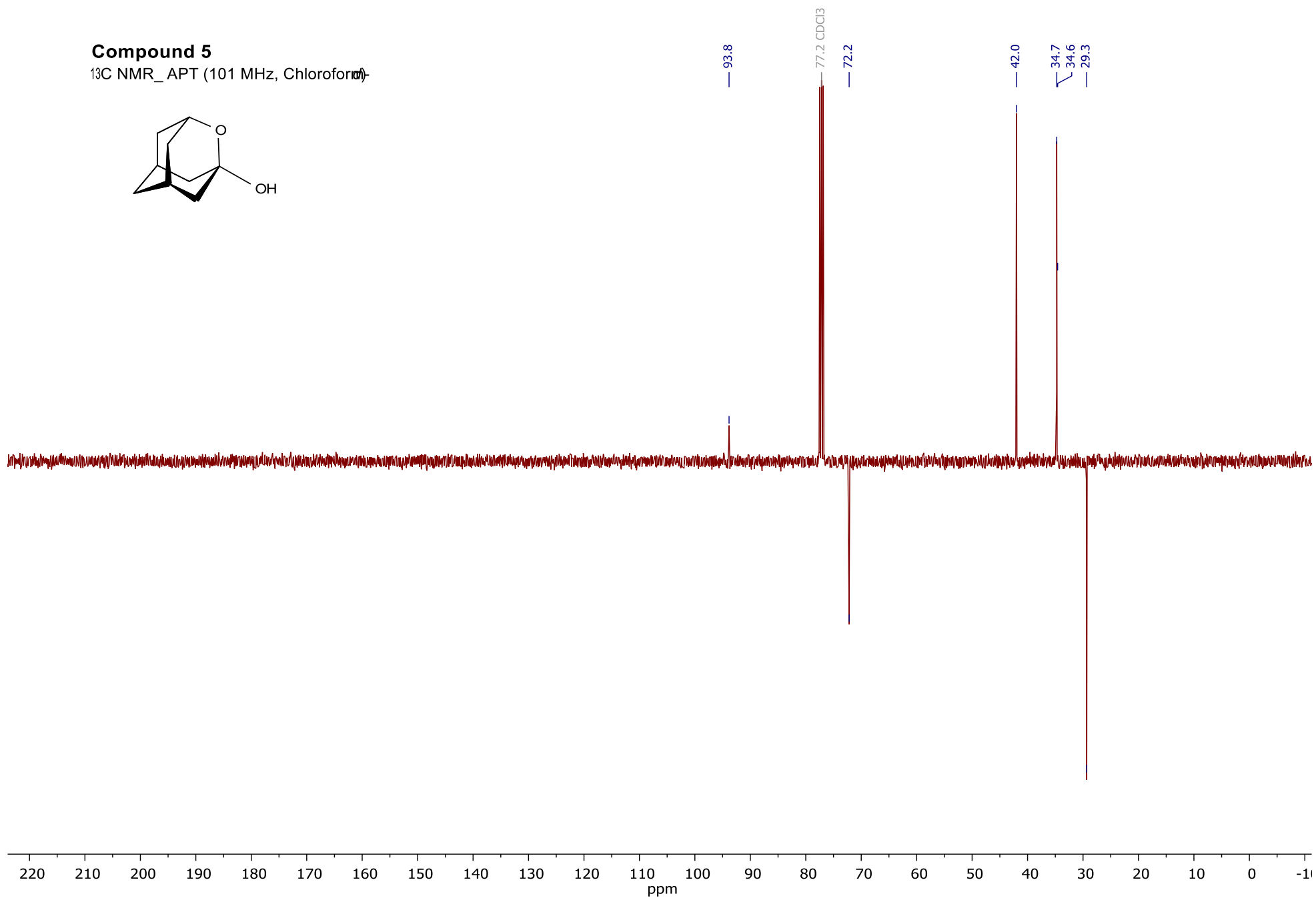
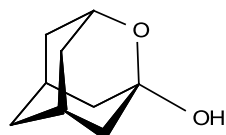
**Compound 5**

<sup>13</sup>C NMR (101 MHz, Chloroform)

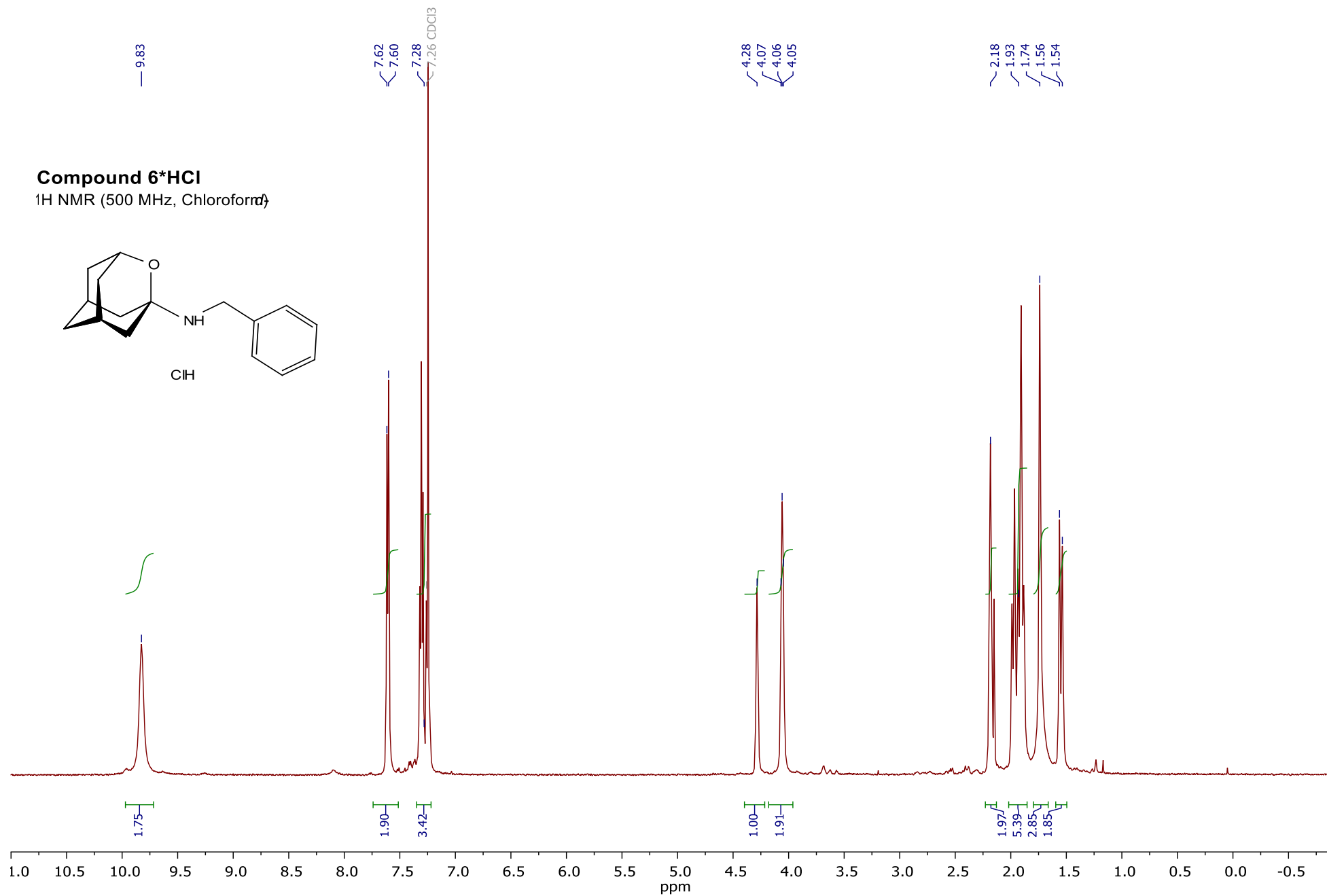
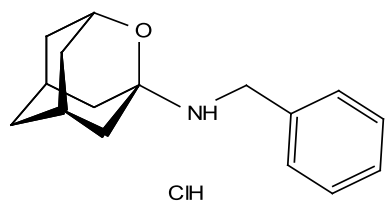


**Compound 5**

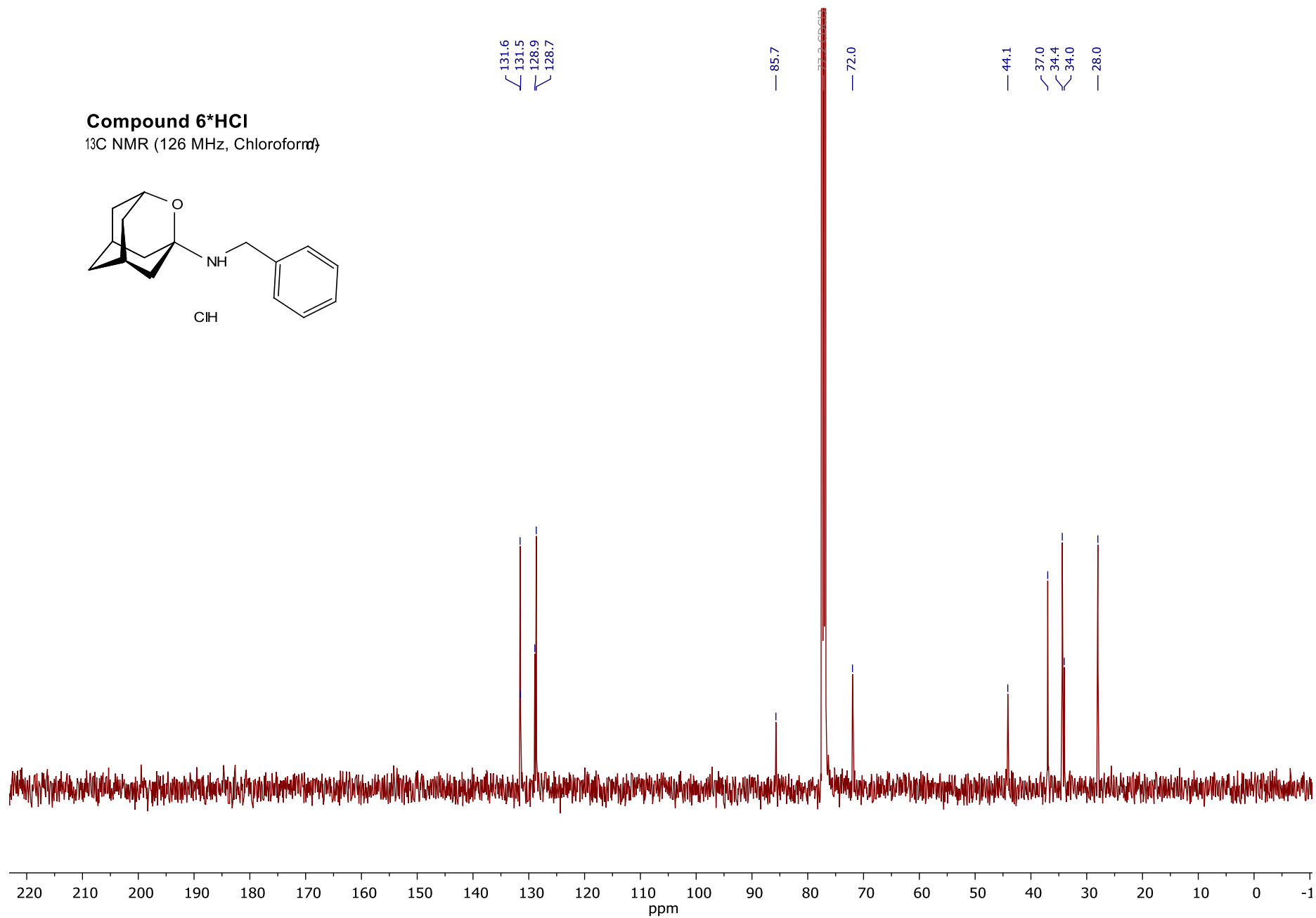
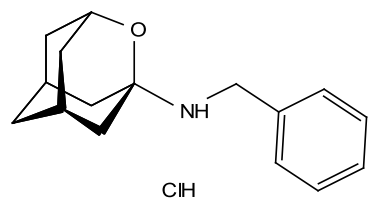
<sup>13</sup>C NMR\_ APT (101 MHz, Chloroform-d)



**Compound 6\*HCl**  
1H NMR (500 MHz, Chloroform-d)

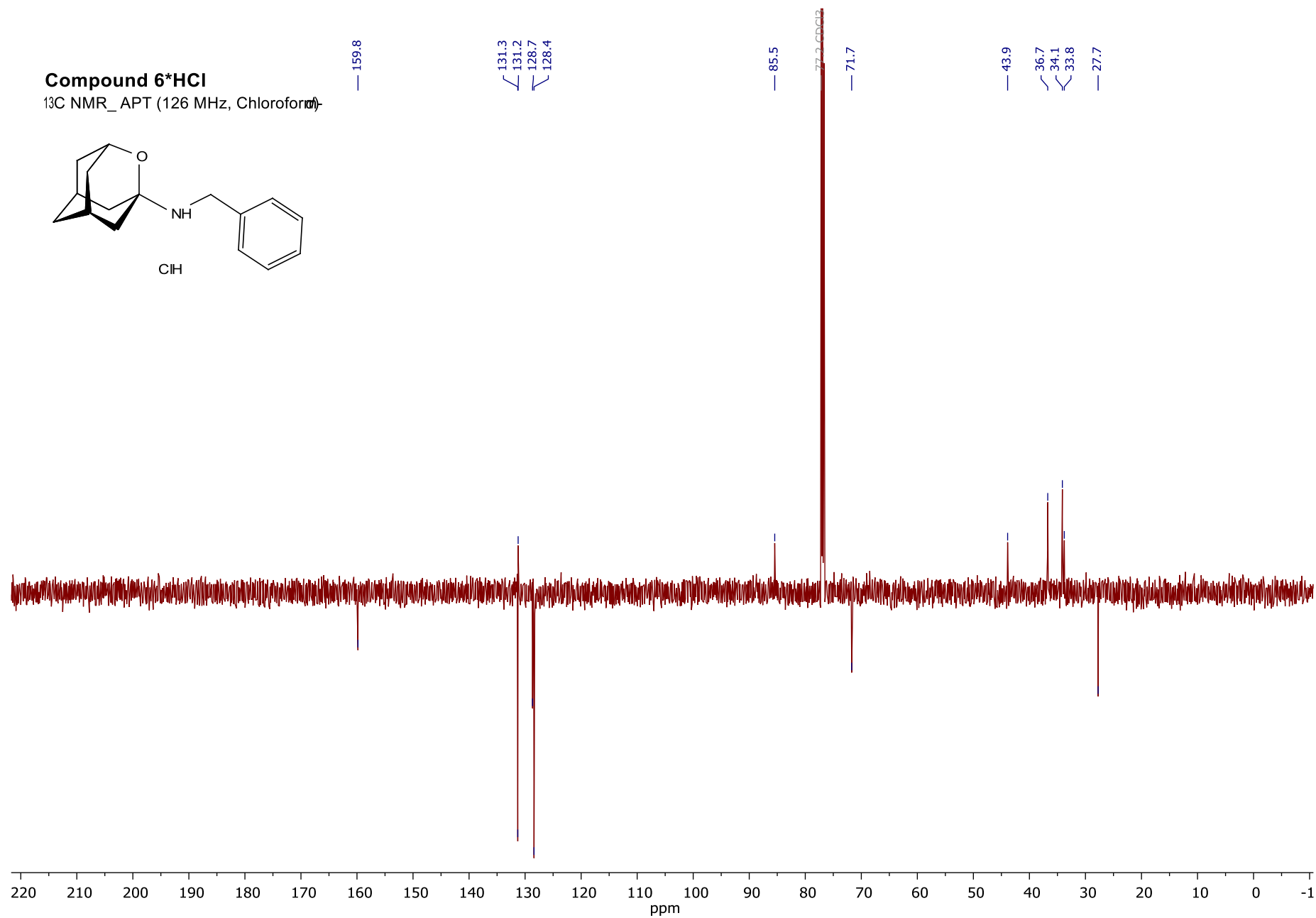
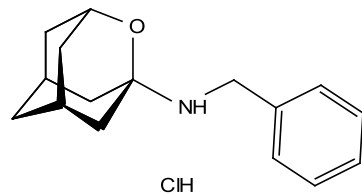


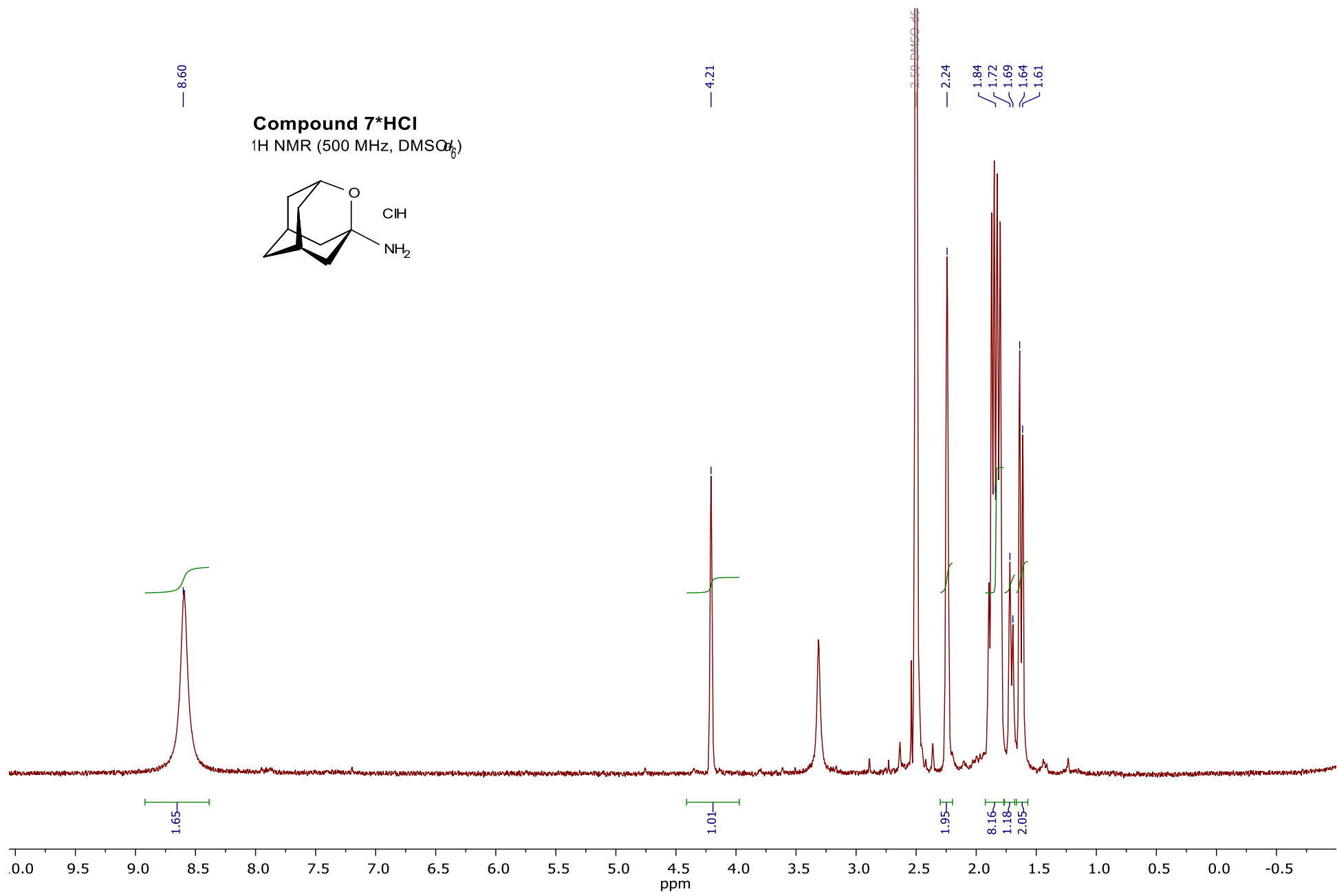
**Compound 6\*HCl**  
<sup>13</sup>C NMR (126 MHz, Chloroform)



**Compound 6\*HCl**

<sup>13</sup>C NMR\_ APT (126 MHz, Chloroform)





**Compound 7\*HCl**

<sup>13</sup>C NMR\_ APT (126 MHz, DMSO-d<sub>6</sub>)

